Letters to the Editor

5,6-Dihydropyridines from isothiocyanates and lithiated 1-(alkylthio)propynes

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The known reactions of organolithium compounds with isothiocyanates result in the formation of thioamides. 1-4

We have established that reactions of lithiated 1-(alkylthio)alkynes with isothiocyanates provide a simple way to prepare not easily accessible 5,6-dihydropyridines. For example, the alkylation of thiolate 2, which is the adduct of lithiated 1-(methylthio)propyne (1) and isopropyl isothiocyanate, results in the previously unknown 2,3-bis(methylthio)-6,6-dimethyl-5,6-dihydropyridine (5), which was isolated in 70% yield (Scheme 1).

Intermediate azatriene 4 was isolated in 96% yield. The structure of compound 5 was confirmed by the data of elemental analysis and IR, ¹H and ¹³C NMR spectra.

The corresponding 5,6-dihydropyridines were also synthesized using 1-(ethylthio)propyne and methyl-, ethyl-, and cyclohexyl isothiocyanates.

Experimental

¹H and ¹³C spectra were recorded on Varian EM-390 (90 MHz, ~20% solution in CCl₄, SiMe₄ as the internal standard) and Bruker AC 300 (¹H: 300 MHz, ¹³C: 75 MHz, ~20% solution in CDCl₃, SiMe₄ as the internal standard) spectrometers. IR spectra were recorded on Specord 75-IR and Perkin-Elmer 283 spectrophotometers in a thin layer. GLC analysis was performed on a Varian 3400 gas chromatograph (with a flame-ionization detector, a 15-m capillary

Scheme 1

$$Bu^{n}Li$$

$$(THF/hexane, -50 °C)$$

$$\longrightarrow [CH_{2}-C \equiv C-SMe \longrightarrow CH_{2} \equiv C=C-SMe] Li^{+} \longrightarrow$$

$$1$$

$$(Me)_{2}CH-N=C=S$$

$$-95 ÷ -35 °C$$

$$(Me)_{2}CH-N$$

$$2$$

$$(Me)_{2}CH-N$$

$$2$$

$$SMe$$

$$H_{2}C=C=C$$

$$(Me)_{2}CH-N$$

$$3$$

$$H_{2}C=C+C$$

$$(Me)_{2}CH-N$$

$$3$$

$$H_{2}C=C+C$$

$$(Me)_{2}CH-N$$

$$SMe$$

$$(Me)_{2}C=N-C$$

$$SMe$$

$$Me$$

$$SMe$$

$$Me$$

$$SMe$$

$$SMe$$

$$SMe$$

$$SMe$$

column with an inner diameter of 0.53 mm and 1.5-µm DB-5 covering, and nitrogen as the carrier-gas).

All procedures were carried out in an atmosphere of N_2 . I-(Methylthio)propyne and isopropyl isocyanate were synthesized by the known methods. 5.6 Tetrahydrofuran was purified using mechanically dispersed KOH (-50 g L⁻¹) and by distillation over LiAlH₄ in the presence of benzophenone in an N_2 atmosphere. n-Butyllithium (1.6 M solution in hexane)- was presented by Chemetall (Germany).

Reaction of lithiated 1-(methylthio)propyne with isopropyl isocyanate and methyl iodide. 1-(Methylthio)propyne (9.1 g. 0.10 mol) was added to a solution of BuⁿLi (0.10 mol) in anhydrous THF (80 mL) and hexane (65 mL) cooled to -60 °C. When the temperature increased to 0 °C, it was decreased to -95 °C, then isopropyl isothiocyanate (10.97 g, 0.11 mmol) was rapidly added. The reaction mixture, whose temperature had increased again to -35 °C, was again cooled to -60 °C, and methyl iodide (22 g, 0.15 mol) was added. The mixture was stirred for 10-15 min at 10-15 °C and diluted with cold water (~60 mL) with intense stirring. The organic layer was separated, the aqueous layer was extracted with pentane (three times) and ether, and the combined organic fractions were dried over K₂CO₃. After the solvent was evaporated on a rotary evaporator, 2-methyl-4,5-di(methylthio)-3-azahepta-2,4,6-triene (4) was obtained (19.34 g, 96.2%) (¹H NMR, δ: 6.10 (d d, 1 H, CH); 5.40 and 4.85 (dd, 2 H, CH₂); 2.27 (s, 3 H, SMe); 2.17 (s, 3 H, SMe); 2.03 (s, 3 H, Me); 1.85 (s, 3 H, Me)) with a minor admixture of azatriene 3 (¹H NMR: 5.14 (s, CH₂)).

After short heating of compound 4, the exothermic reaction of heterocyclization began (the temperature of self-warming was ~200 °C). 2,3-Bis(methylthio)-6,6-dimethyl-5,6-dihydropyridine (5) (14.1 g, 70.1%) with 96.4% purity (GLC)

(b.p. 143-145 °C (15 Torr), n_D^{20} 1.5682) was isolated by distillation in vacuo. 1R, v/cm⁻¹: 600, 670, 800, 810, 870, 880 sh, 940, 950 sh, 990, 1010, 1080, 1130, 1170, 1220, 1250, 1280, 1300, 1350, 1360, 1410, 1440 sh, 1460 sh, 1550, 1590, 2800, 2860, 2910, 2950, 3050, 3100. ¹H NMR (300 MHz, CDCl₃), δ : 5.92 (t, 1 H, H-4); 2.13 (d, 2 H, H-5); 2.14 (s, 3 H, SMe); 2.19 (s, 3 H, SMe); 1.10 (s, 6 H, 2 Me). ¹³C NMR (75 MHz, CDCl₃), δ : 158.52 (C-2); 129.02 (C-3); 128.75 (C-4), 35.80 (C-5); 54.09 (C-6); 16.34 (3-SMe); 12.41 (2-SMe); 28.01 (Me). Found (%): C, 53.63; H, 7.99; N, 6.95; S, 31.52. C₉H₁₅NS₂. Calculated (%): C, 53.73; H, 7.46; 6.96; S, 31.84.

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Nickel acetate-promoted reaction of benzoylcyanamide with 1-aryl-3-methyl-2-pyrazolin-5-ones

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It has been found previously 1.2 that nickel salts and complexes promote the addition of methylene-active β -diketones and esters of β -oxocarboxylic acids to cyanamides. However, similar transformations involving monocarbonyl compounds were unknown. We have established that 3-methyl-2-pyrazolin-5-ones (1a,b), which have a phenyl group with electron-withdrawing substituents at position 1, are also capable of adding to the C \equiv N bond of cyanamides. For example, the corresponding adducts 3a,b, which are new representatives of ketene

aminals of the heterocyclic series, were obtained from compounds 1a,b and benzoylcyanamide (2).

The reaction does not occur in the absence of Ni(OAc)₂. Probably, the nickel enolates, forming from compounds **1a,b** as intermediates, participate in the formation of ketene aminals **3a,b**.

4-(N-Benzoyl)diaminomethylene-3-methyl-1-(4-nitrophenyl)pyrazolin-5-one (3a). A mixture of pyrazolinone 1a (0.219 g, 1 mmol), cyanamide 2 (0.160 g, 1.1 mmol), and Ni(OAc)₂ (0.089 g, 0.5 mmol) in DMF (3 mL) was refluxed

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